	Regi	stration no:		
Total	Nur		HARM	25
	257	2 nd Semester Regular Examination 2015-16 PHARM. ANALYSIS - I Branch: Pharmacy Time: 3 Hours Max Marks: 100 QUESTION CODE: W508 Answer Part-A which is compulsory and any four from the Part-B. The figures in the right hand margin indicate marks.	PH201	25
		Part-A (Answer all the questions)		
Q.1		·	2 x 10)	
	a)	The difference between equivalence point and end point of titration is called as 257 257		25
	b)	The pH of 0.01N nitric acid is		
	c)	is used as a coating agent in Volhard's method of		
		titration.		
	d)	Two indicators used in acid-base titration are and		
	e)	Eriochrome black T and murexide are used as indicators in method of titration.		25
	f)	Examples of two organic precipitants are and		
	g)	0.0253g contains significant figures.		
	h)	The solution which resists the change in pH even after addition of small amount of acid or base is called		
	257 i)	In titration, the solution which is generally taken in a burette is called and the substance which is titrated is called		25
	j)	Balance the redox reaction: $Cr_2O_7^{-2} + Fe^{+2} \rightarrow Fe^{+3} + Cr^{+3}$		
Q.2		Answer the following	2x10)	
	a)	Define and differentiate between accuracy and precision.		
	b)	Classify different types of errors and give one example from each category.		25
	c)	What is Universal indicator?		
	d)	Define Lewis acid and base. Give at least one example from each.		
	e)	What is common ion effect?		
	f)	Give the principle of Fajan's method of titration.	7	age.

	9)	What is self indicator? Give example.		
	h)	Why acetic anhydride is used in the preparation of 0.1N Perchloric acid? 257		257
	i)	What are Argentometric titrations?		
	j)	What are ligands? Classify them with examples.		
		Part-B (Answer any four questions)		
Q.3	a)	Write the steps for minimization of errors in pharmaceutical analysis.	(7)	
	b)	What do you mean by neutralization curve? Mention its significance by taking the example of titration of a strong acid against a strong base.	(8)	257
Q.4	a)	Define and describe different types of complexometric titrations.	(6)	
	b)	Write a note on chelating agents.	(3)	
	c)	Give the procedure for determination of calcium by EDTA titration.	(6)	
	257	257 257 257 257 257 257		257
Q.5.	a)	What are the advantages of non-aqueous titrations?	(2)	
	b)	Give a note on the solvents used in non-aqueous method of titrations.	(6)	
	c)	Write the method of preparation and standardization of 0.1N Potassium methoxide.	(7)	
Q.6	a)	Write down briefly the steps involved in gravimetric analysis by precipitation methods.	(8)	257
	b)	What is thermogravimetric curve? Mention its features. Mention the application with suitable example.	(7)	
Q.7.	a)	Give the principle behind Mohr's method of titrations.	(6)	
	b)	Explain the principle and procedure behind estimation of sodium chloride by Volhard's method of titration.	(9)	257
Q.8	a)	Write the theory of redox titrations.	(3)	
	b)	How will you calculate strength and equivalent weights of oxidizing and reducing agents? Explain with suitable examples.	(6)	
Q.9	c)	Write a note on preparation and standardization of 0.1N KMnO ₄ solution. ²⁵⁷ Answer the following (Any three)	(6) (5x3)	257
	a)	Masking and demasking in complexometric titration.		
	b)	Ideal characteristics of primary standard		
	c)	Quinonoid theory of indicators		
	d)	Law of mass action 257 257 257 257 257 257		Page 2

Regist	ration	no.			

Total Number of Pages: 2

B.PHARM

2ND Semester Back Examination 2015-16 PHARMACEUTICAL ANALYSIS - I

BRANCH: <u>B.Pharm.</u> Time: 3 Hours Max Marks: 70 Q.CODE:W509

Answer Question No.1 which is compulsory and any five from the rest.

The figures in the right hand margin indicate marks.

Q1 Answer the following questions:

(2 x 10)

- a) What is Self-indicator? Give example.
- b) Define normality and molarity.
- c) What is solubility product?
- d) Distinguish between Accuracy and Precision.
- e) Explain Common ion effect.
- **f)** Write the Ideal characteristics of primary standards.
- g) Define masking and demasking agents with examples.
- **h)** Write the equation of Henderson-Hasselbalch equation.
- i) Calculate the PH of 0.01N of HCl.
- j) Name two indicators used in Complexmetric titrations.

Q2 a) Write briefly on Law of mass action.

(5x2)

- b) Define Neutralization curve of strong acid and strong base.
- **Q3** a) Discuss briefly on Theories of Indicator.

(5x2)

- b)Write the preparation and standardization of 0.1N TBAH
- Q4 Define Non aqueous titration. Explain the methods to estimate any acidic drug by Non-aqueous method of titration. Indicators used in Non-aqueous titration.

(10)

Q5		(5x2)	2				
Q6		Write short note a)Organic prec b)Co-precipitat	ipitants.			(5x2)	2
Q7	a) b)	Volhards metho	d.		tion of chlorides b		
Q8	a) b) c) d)	Write short note Amino acid titrat Pharmaceutical Buffer solution. Universal indicat	cion. error.			(5 x 2)	2
257		257	257	257	257	257	2
257		257	257	257	257	257	2

Reg	istra	ation No:	
Tota	al Nu	umber of Pages: 02 257 257 257 257 <u>257</u>	B.PHARM 15PH203
		2 nd Semester Regular Examination 2015-16 HUMAN ANATOMY & PHYSIOLOGY-II BRANCH: Pharmacy Time: 3 Hours	
257		257 Max Marks: 100 257 257 Q.CODE: W477 Answer Part-A which is compulsory and any four from Part-l	25 B.
		The figures in the right hand margin indicate marks.	
Q1 257	a)	Part – A (Answer all the questions) Answer the following questions: Gastrin secretion is stimulated by	(2 x 10)
	b)	In the lung alveoli P _A CO ₂ is about mmHg and diffusibility of	
	•	CO_2 is (higher/lower) than O_2 .	
	c)	GFR per day in young healthy adult is about litres and daily urine	
		output is litres.	
257	d)	Normally colloidal osmotic tension in the glomerular capillary is about	25
		mmHg and the effective filtration pressure is about	
	e)	Rhodopsin is made up of a combination of &	
	f)	The neurotransmitter released at the postganglionic sympathetic nerve ending	
		is and parasympathetic nerve ending is	
257	g)	GnRH stimulates the secretion of &	25
	h)	Insulingluconeogenesis and lipolysis.	
		(Inhibits/promotes).	
	i)	ADH in human being is	
	j)	A myelinated nerve fibre conducts nerve impulse than non-	
257		myelinated nerve fibre. (faster/slower)	25
Q2	a)	Answer the following questions: <i>Short answer type</i> Name the hormones released from adrenal cortex.	(2 x 10)
	b)	What is chloride shift?	
257	c)	Define tidal volume.	

d) What is peptic ulcer?

	e)	Define diabetes mellitus.	
	f)	What is Myxedema?	
257	g)	Name any two excitatory neurotransmitters released in CNS.	
	h)	Define digestion.	
	i)	What is colour blindness?	
	j)	Define action potential?	
Q3 ₇	a)	Part – B (Answer any four questions) Describe the anatomy & physiology of liver. Add a note on its function.	(10)
-	b)	Write about the location and functions of different types of taste buds.	(5)
~ 4	,	••	` ,
Q4	a)	Explain briefly about the mechanism of respiration.	(10)
	b)	Write a note on voice box.	(5)
→ 257	٠,	257	(40)
Q5	a)	Write in detail about the physiology of urine formation.	(10)
	b)	Describe EEG with the nature of its waves.	(5)
Q6	a)	Enumerate in detail about the steps involved in the neurohumoral transmission	(10)
		in parasympathetic nervous system.	
257	b)	Write a note on reflex arc.	(5)
Q7	a)	Write in detail about the physiology of menstruation.	(10)
	b)	Write a note on spermatogenesis.	(5)
~	•	D: 0	
Q8	a)	Briefly enumerate about the physiology of hearing.	(10)
257	b)	Name the hormones released from thyroid glands along with their functions. 257 257 257 257 257 257	(5)
Q9	a)	Describe about the structure and functions of spinal cord.	(10)
	b)	Draw a neat labeled diagram of human eye.	(5)

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257	nsw	ver Questic	on N	ium.	AN A	NAT BRAN Tir Ma Q.C Ch is	OMY NCH: ne: (x Ma COD	/ & P Pha B Ho arks E: W	rmac ours : 70 /478	iOLC y y an	OGY-	II ny fi	ve fro		PH.2.3 ne rest.
Q1		The fig Write briefly	_			rıgr	it na	nd i	març	gin i	ndic	ate	mark	S.	(2 x 10)
	a)	Gastric Juice			C										,
	b)	Tidal volume	and v	vital c	apaci	ity									
257	c)	Chloride shif	ì		257			257			257			257	2
	d)	Ultra-filtratio	n												
	e)	Middle ear ca	avity												
	f)	Write about h	numar	ı Utei	us.										
257	g)	Write about to	he hu	man p	pancre	eas.		257			257			257	2
	h)	Cerebellum													
	i)	Liver													
	j)	Parathyroid g	land												
Q2 ⁷		What is ANS describe the a		•										and 7	(10)
Q3		Describe the	huma	n dige	estive	syste	m wi	th a n	eat an	ıd lab	eled d	liagra	m. Wr	ite a	(10)
		note on diges	tion o	of foo	d in s	mall i	ntesti	ne.							
Q4		Describe the	variou	us par	t of u	rinary	/ tract	syste	em wi	th a n	eat ar	nd lab	eled		(10)
257		diagram. Brie	efly m	entio	n abo	ut uri	nary t	racti	nfecti	on.	257			257	2
Q5		What is mens	struati	on? V	What a	are th	e vari	ous s	tages	of me	enstru	al cyc	ele?		(10)
Q6		Describe the			-		abou	t the	senso	ry and	d mot	or co	tex and	d	(10)

Q7 257 Q8		on commonest type Answer any two of	es of refractory erro		n eye. Write a brie	257	(10) (5 x 2)
	a)	Oogenesis					
	b)	Spermatogenesis					
257	c)	Pregnancy and its n	naintenance.	257	257	257	251
	d)	Answer any two of	the following:				
257		257	257	257	257	257	251
257		257	257	257	257	257	257
257		257	257	257	257	257	257

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Tota	ıl Nu	ımber of Paç	ges: 02	2	257			257			25	7		257	B.PHARN 15PH20	_
257		2 Answer Pa The fi		/hio	PH E 257 I	IARI BRAI Tir Max Q.C s co	MAC NCH: ne: (Ma COD mpu	OGI Pha 3 Ho Irks E: W	NOS rmad urs : 10 /566 ry ar	Y-I y 0 nd a	ny f	our f	from		-B.	
Q1 ₂₅₇	a)	Answer the c	orrect o	Part	<u>t – A</u> n of	(An the fo	<u>swer</u>	all tl			<u>ons)</u>			257	(2 x 10)	
	b)	Unit of micro	•		1100					(M:		er/Mi				
	c)	Drug having	•	latil	e oil	and f	ixed o	oil.		`		a/Nutr				
	d)	Family of Lin	nseed oi	il.						(Lil	iacea	e/Lina	iceae)			
	e)	Nitrogen is so	econdar	y nu	trien	ıt.					(7	Γrue/F	alse)			
257	f)	Chemical res	ponsible	e for	yell	ow co	olour	of tur	meric	. (Cu	rcum	inoids	/Jalapi	in) 257		
	g)	Myrrh is an O	Oleo-gu	m re	sin.						(Tı	rue/Fa	lse)			
	h)	Family of Be	es wax.							(B	ovida	ie/Api	dae)			
	i)	Pectin is a re-	versible	col	loid.						(Tr	ue/Fal	se)			
257	j)	Datura is hav	ing whi	ich t	ype (of Sto	mata.	257	(A	nisoc	ytic/]	Paracy	rtic)	257		
Q2	a)	Answer the Who coined													(2 x 10)	
	b)	What is Seed	Propag	gatio	n?											
	c)	Define Stoma	atal nun	nber	•											
257	d)	Write two fur	nctions	of C	ytok	inin.		257			25	7		257		
	e)	Write the che	emical c	onst	ituer	nts of	Agar									

f)

h)

i)

j)

What is Polyploidy?

What is Vein-islet number?

Define Optical rotation.

Write the biological source of Cod liver oil.

Write the biological source of Turmeric.

Part - B (Answer any four questions) Write in details about the classification of Crude drugs with suitable examples. Q3 (10)b) Write the difference between Organized and Unorganized drug. (5) Write the various conditions of adulteration. Q4 (5) Describe in details about the different types of adulterants with examples. (10)Describe the different methods of Pest control management. (10)Write a note on Soil. (5) Q6 Describe the detail and systematic Pharmacognostic study of Honey. (10)Write the Biological source, Chemical constitution and Uses of Asafoetida. (5) Write the applications of Auxins and Gibberellin. (10)Q7 Define Mutation with it's detail classifications. (5) Write details about Secondary metabolites of plant origin with their properties. Q8 (10)Write the Biological source, Chemical constitution and Uses of Lard. (5) Q9 Write the Biological source, Chemical constitution and Uses of Wool fat and (10)Gambir. Write the classification of tannins with suitable examples. (5)

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257		257	NI	_ 4 .	257	Q.C		E: W	/565		257		.a fua	257	
An	ISW	er Question The fig						-		-		-			ie rest.
Q1		Answer the fo	ollow	ing qı	ıestio	ns:									(2 x 10)
257	a) b) c) d) e) f) g) h) i)	Define and cl What are resi Goldbeater's Write the bio Define fibres Name any for Write the che Define Pharm Write the bio Write the bio	assifyns? G skin t logica . Writ ur alle mical naceur logica	Tannetive tweest. It sould be the ergense constical and sould sould be the constical and sould be the tree that the tree tree the tree tree tree tree	nins wo ex rce ar biolo s of na tituer uids ar rce ar	ample ample gical atural ats and md wr nd che	emica sourc original d use ite its emica	l cons e of s n. of Le vario	stituerilk. mon gous so	grass. ources. outs of	Asafo	amon.		257	(= 1.10)
Q2		Explain in de volatile oil.	tail al	out t	he pro	operti	es and	d gen	eral n	nethod	l of ex	xtracti	ion of		(10)
	a) b)	Write details Write notes o Photosensitiz Bentonite	n the	follo	_		profile	e and	its im	ıportaı	nce o	f clov	e.	257	(5+5) (5) (5)
Q5	·	What are re biological sou	arce,c	hemi	•			_							(2+3+5)
Q7 ²⁵⁷ Q8	a) b) a)	Write short no Natural Aller Antioxidants Write the Pha Mention the b Podophyllum	gens armac piolog	ognos				257 cons 1	tituen	ts and	uses			257	(5) (5) (10) (2.5x 4)
	b) c) d)	Myrobalan Mentha Saffron	•												

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PHARMA Answer Se		CAL C BI	HEMIS RANCH Time Max I Q.CO	TRY-I : PHA e: 3 Ho marks DE: V oulsory	II (OR RMA ours :100 V437 y and a	RGAN CY any F	IIC C	THEM	Section 'B'.	257 257
			Se	ction A	4					
Q1. Answer the fol	lowing q	uestion	s:						(2 x 10)	
a) Hybridisation of 0	Carbon ir	thene	e is:							
257 i) sp 25	₅₇ ii) sp ²	25	viii) sp ³		iv) sp	3d		257	257	257
b) The quantum num	nber which	ch gives	idea abo	out the	orbita	ls is:				
i) Principal (Q.N.	ii) Aziı	muthal Ç).N.	iii) M	agnet	ic Q.1	N.	iv) both ii & iii	
c) The order of react	tivity of a	alkyl hal	lides tow	ards S	N ¹ rea	ction	is:			
i) $3^0 > 2^0 > 1^0$.	7	ii) 1 ⁰ >2	$2^{0} > 3^{0}$		iii) 3 ⁰	$>2^0=1$	0	257	iv) $3^0 = 2^0 > 1^0$	
d) The anti -Markon					201			257	201	201
i) Carbocatio	n mecha	nism	j	ii) Free	e radic	al med	chanis	sm		
iii) Rearrang	gement		j	iv) Res	sonanc	e.				
e) Alkanes are know	n as para	ıffin due	e to their	:						
i) Less reacti	vity	ii) Hi	gh polari	ity	²iii) H	ligh vi	iscosi	ty	iv) More reactivity	257
f) The basic nature of	of amines	are in t	he order	of:						
i) 3 ⁰ >2 ⁰ >1 ⁰ i	n gaseou	s phase	ii) 2 ⁰ >3 ⁰	$^{0}>1^{0}$	iii) 1 ⁰	$>2^0>3$	3^0		iv) Bothi& ii	
g)The electronic cor	nfiguratio	on of an	element	is $1s^2$ 2	$2s^22p^63$	s^23p^6	3d ⁵ 4s	¹ .This	represents its:	
i) Excited sta	ite	ii) Gro	und state	ed	ziii) C	ationi	ic for	n ,	iv) Anionic form	257
h) High boiling poin	t of wate	er is resu	ılt of:							
i) Intermolec	ular hyd	rogen bo	onding	ii) Intra	amoleo	cular l	ıydro	gen b	onding	
iii) Bothi& i	i		j	iv) Hig	gh spec	ific h	eat			
i)Alkyl halide can b	e convert	ed into	alkene b	y:	257			257	257	257

i) Nucleophilic substitution reaction ii) Elimination reaction

- iii) Addition reaction
- iv) Rearrangement reaction

j)In which of the following molecules, the central atom does not have sp³ hybridization?

- i) CH
- ii) SF₄
- iii) BF₄
- iv) NH₄⁺

Q2 Answer the following questions:

 (2×10)

- a) Explain Wurtz Reaction. Can methane be prepared by this method?
- b) What are gem-dihalides and vicinal-dihalides? Give examples.
- c) Write Williamson's synthesis.
- d) Write the IUPAC name of

$$CH_{3} \xrightarrow{\qquad CH \qquad \qquad CH \qquad \qquad CH_{3}} \xrightarrow{\qquad CH_{3}} CH_{3}$$

- e) What are the isomers possible for formula C₂H₆O? Write their IUPAC name.
- f) What is Lindlar's catalyst? What is its use?
- g) What are nucleophiles? Give examples.
- h) Explain Cannizaro reaction.
- i) What is Saytzeff rule? Explain with example.
- j) Explain Diel's Alder Reaction.

Section B

Q3 a) Explain the Bohr's Atomic theory and its limitations.

(10)

b) Differentiate between orbits and orbitals.

- **(5)**
- Q4 a) Define and classify chemical bond. Explain covalent bond and their types with examples. (3+7)
 - b) Explain Hydrogen Bonding.

(5)

- Q5 a) Define Alkanes with examples. Write any four method of preparation and anyfour
 - chemical properties of Alkanes.

(2+8)

b) Explain Baeyer strain theory.

(5)

Q6 a) Explain SN² mechanism with suitable example.

(10)

b) Differentiate between Inductive & Electrometric effect.

(5)

Q/a)	Define Alkenes with e	xampies. Write	any four method	i or preparation and a	illy loui
	chemical properties o	f Alkenes.			(2+8)
² 57)	Explain the acidic pro	perty of actylen	ic hydrogen. Giv	re reactions due to th	e acidic
	character.				(5)
Q8 a)	Define and classify Ca	arboxylic acids?	Write any four i	methodof preparation	ı of
	Carboxylic acid. Expl	lain the effect of	f substituent on a	icidity.	(3+4+3)
₂₅ b)	Explain the Free radic	al substitution r	eaction mechanis	sm with example.	₂₅₇ (5)
Q9 a)	Define and classify Ar	mines? Write an	y four method o	f preparation and any	yfour
	chemical properties.				(2+8)
b)	How you can distingu	ish primary, sec	ondary and tertia	ary alcohol	(5)
257	257	257	257	257	257
257	257	257	257	257	257
257	257	257	257	257	257

Reg	istra	ation no:							
Total Number of Pages: 02 257 257 257 257 257 B.Pharm PH.2.10									
2 nd Semester Regular / Back Examination 2015-16 PHARMACEUTICAL CHEMISTRY –II BRANCH: Pharmacy									
257		Time: 3 Hours 257 Max Marks: 70 257 25 Q.CODE: W438			2				
Α	nsv	ver Question No.1 which is compulsory and any five from The figures in the right hand margin indicate marks.	the re	∍st.					
Q1 257	a) b) c) d) e) f) g) h) i)	Answer the following questions: What is Diels Alder Reaction? Define Markovnikor rule with suitable examples. Describe saytzeff rule with suitable example. What are the bond angles in SP3, SP2 and SP hybrid orbitals? Differentiate inter and intra molecular hydrogen bonding with examples. Write IUPA name of CH ₃ - CH ₂ - CH-CH ₂ = CH ₂ 257 CH ₂ - CH ₂ - CH ₃ Why acetylene is acidic? Write structure of the following compounds. Write two form for Electrophilic reagents and Nucleophilic reagents. What happens when Diethyl ether reacts with PCl5.	· ·	x 10)	2				
Q2 ₂₅₇	a) b)	Discuss various Postulates of Bohr's atomic structure & write its limitation Write notes on chemical bonding.	7	(5) (5)	2				
Q3	a)	What is hybridization? Discuss the types of hybridization with appropriate examples.	((5)					
	b)	Discuss Baeyer strain theory and its limitation.	((5)					
Q4 257		Outline any two general methods of preparation for Alkenes and Alkynes. Discuss the mechanism of free redical substitution reactions of alkane.	(*	10)	2				
Q5	a) b)	Write notes on following SN1 and SN2 reactions. Outline any four general method of preparation of alcohols.		(5) (5)					

Q6 Q7	a) b) a) b)	 Write notes on Williamson's Synthesis. Write any two general methods of preparation and any two chemical properties of Cycloalkanes. 							
Q8 257	a) b) c) d)	Write short notes on : Chemical Bonding Free Radicals Basicity of Amines Nucleophilic Substituti	on reaction.	257	257	(2	2.5 x 4)		
257		257	257	257	257	257	257		
257		257	257	257	257	257	257		
257		257	257	257	257	257	257		
257		257	257	257	257	257	257		

Regi	stra	ation No:		
Tota	l Nu	imber of Pages: 02 257 257 257 257 257 2	B.PHARM 15PH209	25
		2 nd Semester Regular Examination 2015-16 BUSINESS COMMUNICATION IN ENGLISH BRANCH: Pharmacy		
		Time: 3 Hours		
257		²⁵⁷ Max Marks: 100 ²⁵⁷ 257		25
		Q.CODE: W273		
	A	Answer Part-A which is compulsory and any four from Part-	В.	
		The figures in the right hand margin indicate marks.		
		Part - A (Answer all the questions)		
Q1 ₂₅₇		Answer the following questions:	(2 x 10)	25
	a)	Since communication is mostly through language, communication skills are		
	b)	sometimes described as The reader's ability to read between lines refers to		
	c)	Carrying out orders, direct or indirect appeals, requests for help etc. by the		
	•	subordinates are the forms of		
	d)	The process of searching a text to have a specific piece of information is called		
257	e)	The ideas generated through brain-storming are placed in different positions to		25
	f)	form a diagram is called as A written document that analyses and presents factual information, comments,		
	٠,	suggestions, recommendations etc. is known as		
	g)	The tendency to think that one's own culture is superior to other cultures is		
	ل ـ ۱	known as communication.		
	h)	is an example of message		
257	i)	is a procedure used by organisations to assess the candidate's		25
	•	technical knowledge, skills, personality traits, level of commitment, basic		
		interests etc.		
	j)	Being a team member, while you are communicating with your team mates in		
		an organisation, the communicating category is known as		
Q2		Answer the following questions: Short answer type	(2 x 10)	
257	a)	What is Grapevine? 257 257 257		25
	p)	Write any two barriers to Cross-cultural communication. What is Diagonal Communication?		
	c) d)	Make the following sentences Bias free:		
	,	i)BhimaBhoi was blind.		
	_	ii)Mrs.Tripathy is a pharmacist		
	e)	Write any two examples of Informal communication.		

What is Coherence?

g) What is Global comprehension?

f)

257	257	h) i) j)	While two persons are communicating each other with which words/phrase, will you attract the attention of the person whom you need to interact? What is BCC? Explain its usage. Define Team and Group.		257
257	Q3	a) b)	Part – B (Answer any four questions) Discuss the pros and cons of Speaking and Writing. Narrate "how a Chronological CV is different from a Functional CV".	(10) (5)	257
	Q4	a)	As Marketing Manager of a Pharmaceutical company, draft a congratulatory e-mail message to the managers of different departments (such as Quality Assurance & Research and Production) for successful production of a new vaccine which has more demand in present pharmaceutical market.	(10)	
		b)	Contrast between Group Discussion and Debate.	(5)	
257	Q5	a)	Imagine that you are the Production Analyst of a Pharmaceutical company. The Managing Director of your company has communicated you to find out what types of effort would be given to have an ample production. Draft a report.	(10)	257
		b)	What is a Topic Sentence?	(5)	
257	Q6	a) b)	Discuss the different stages of Team formation. "Speech is Context-dependent". Justify. 257 257 257	(10) (5)	257
	Q7	a)	Leadership skills are required for the growth of a business organisation. Briefly discuss the principles of leadership that helps in the development of an organization.	(10)	
		b)	Write a short note on Soft Skills.	(5)	
257	Q8 ₂₅₇	a) b)	Write down the differences between Formal and Informal communication. What is Brain-storming?	(10) (5)	257
	Q9	a)	Being a Pharmaceutical Sales Manager, how will you give a briefing of a new Pharmaproduct lunched recently by your company. Give an oral presentation to a group of Medical practitioners.	(10)	
		b)	What is Cohesion? Explain with example.	(5)	
257	257		257 257 257 257 257		257

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Total Number of Pages: 02

B.PHARM 15PH211

2nd Semester Regular Examination 2015-16 Hospital and Community Pharmacy BRANCH: Pharmacy

Time: 3 Hours
Max Marks: 100
Q.CODE: W601

Answer Section 'A' which is compulsory and any Four from Section 'B'. The figures in the right hand margin indicate marks.

		SECTION- A		
Q1	- \	Answer the following questions:	(2 x 10)	
257	a) b) c)	Large Hospital have numbers of beds is used for Moist heat sterilization method is used for distribution of Charged floor stock drugs.		
257	d) e) f) g) h) i)	Large volume paranterals contains preservatives. True / False Objectives of PTC are and Terminal method of family planning are and In inventory control EOQ stands for Gaseous sterilization is done by using gas. Full form of HEPA is 257 257 Objectives of PTC are and		
Q2 257	a) b) c) d) e) f) g) h) i)	Mention the temperature ranges for different storage conditions. What do you mean by buffer stock? Mention three materials whish are sterilized by Autoclave. What are surgical cotton? Composition of Universal antidote. 257 257 Classify hospitals according to ownership. Mention the advantages of floor stock system. Composition of Infection control committee. Define OTC drugs. Master formula card.	(2 x 10)	
		SECTION- B (answer any four)	(40)	
Q3	a)	Define Hospital. Classify them and note down the functions of Hospitals.	(10)	
	b)	Define and write the functions of Hospital pharmacy.	(5)	
Q4	a)	Define Inventory control and describe the techniques of Inventory control.	(10)	
	b)	Write notes on Purchase procedure of drugs.	(5)	
Q5	-	Describe the Temporary methods of family planning. 257 Write in brief about retail drug store.	(10) (5)	

Q6	a)	Mention the different methods of drug distribution to in-patients, describe floor stock system.	(10)
	b)	Write notes on surgical cotton and adhesive tapes	(5)
Q 7	a)	Define sterilization. Describe the different methods of sterilization with suitable examples	(10)
	b)	Write about the emergency treatment of poisoning.	(5)
Q8	a) b)	Describe elaborately on Pharmacy and Therapeutic committee. Write about Infection control committee.	(10) (5)
²⁵⁷ Q9	a) b)	Describe in details about Hospital formulary system. Write in brief about Ethics committee.	(10) (5)

Reg	istra	ation no: 257 257 257 257 257	25
Tota	al Nu	umber of Pages: 01 B.F	HARM PH.2.9
257		257 2 nd Semester Back Examination 2015-16 HOSPITAL PHARMACY BRANCH: Pharmacy Time: 3 Hours Max Marks: 70 Q.CODE: W602	28
2 A	nsv	wer Question No.1 which is compulsory and any five from the i	est. 25
Q1	a) b) c) d) e) f)	Mention the number of pharmacist required as per bed strength of a hospital. Note the objectives of PTC. What is ABC analysis? Define Hospital Formulary system. What is satellite Pharmacy? Write about various types of storage conditions.	2 x 10)
	g) h) i) j)	What are large volume parenteral? Mention the advantages of Unit dose dispensing What do you mean by ambulatory patient? Mention the abilities required of a hospital pharmacist.	
Q2 ²⁵⁷	a) b)	Define and describe the functions of Hospital pharmacy. Define and classify Hospital	(5) (5)
Q3	a) b)	Describe the contents of Formulary. Define Drug Information centre. Describe different sources of information.	(5) (5)

Define PTC. Describe its function, composition and role.

Write notes on Unit dose dispensing system of drug distribution.

Mention the different methods of sterilizations with suitable example of

Describe the Floor stock system of drug distribution

Q6 a) Write notes on Inventory control.

b) Purchase procedure of drugs

materials sterilized by them.

Write short notes on any two:

a) Patient medication profile

c) Adhesive tapes

Pyrogen test

b) Describe rhe merhod of moist heat sterilisarion

b) Functions of Central sterile supply department

d) Responsibilities of Hospital Pharmacist

(10)

(5)

(5)

(5)

(5)

(5)

 (5×2)

Q4

Q7

Q8

Q5 a)

b)

a)